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2-Methylsulfanyl-9H-1,3,4-thiadiazolo-[2,3-b]quinazolin-9-one

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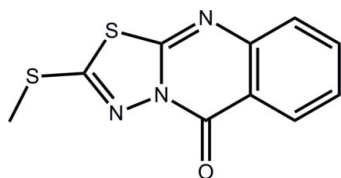
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Key indicators: single-crystal X-ray study; $T = 100$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.029; wR factor = 0.082; data-to-parameter ratio = 14.6.

In the title compound, $\text{C}_{10}\text{H}_7\text{N}_3\text{OS}_2$, the 16 non-H atoms are almost planar (r.m.s. deviation = 0.037 Å) and the S-bound methyl group is *syn* to the ketone O atom. In the crystal, centrosymmetrically related molecules are connected by pairs of $\text{C}-\text{H}\cdots\text{O}$ interactions between the ketone O and methyl H atoms. The dimeric aggregates are connected into a linear supramolecular chain along the b axis *via* $\pi-\pi$ interactions occurring between the five-membered and benzene rings [centroid-centroid distance = 3.6123 (9) Å]. The chains assemble into layers in the bc plane *via* $\text{S}\cdots\text{S}$ interactions involving the endocyclic S atoms [$\text{S}\cdots\text{S} = 3.4607$ (6) and 3.4792 (6) Å].

Related literature

For recent studies on the synthesis and biological properties of quinazoline-4(3H)-one derivatives, see: El-Azab & ElTahir (2012); El-Azab *et al.* (2011). For the synthesis and antimicrobial activity of the title compound, see: El-Azab (2007).



Experimental

Crystal data

 $\text{C}_{10}\text{H}_7\text{N}_3\text{OS}_2$ $M_r = 249.31$

Monoclinic, $P2_1/c$
 $a = 11.8193$ (4) Å
 $b = 4.9841$ (2) Å
 $c = 17.4985$ (6) Å
 $\beta = 91.453$ (3)°
 $V = 1030.48$ (6) Å³

$Z = 4$
 Cu $K\alpha$ radiation
 $\mu = 4.53$ mm⁻¹
 $T = 100$ K
 $0.30 \times 0.10 \times 0.03$ mm

Data collection

Agilent SuperNova Dual diffractometer with an Atlas detector
 Absorption correction: multi-scan (*CrysAlis PRO*; Agilent, 2012)
 $T_{\min} = 0.344$, $T_{\max} = 0.876$

3781 measured reflections
 2110 independent reflections
 1937 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.018$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.029$
 $wR(F^2) = 0.082$
 $S = 1.09$
 2110 reflections

145 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.36$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.27$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{C}10-\text{H}10A\cdots\text{O}1^i$	0.98	2.32	3.170 (2)	145

Symmetry code: (i) $-x + 1, -y + 2, -z + 1$.

Data collection: *CrysAlis PRO* (Agilent, 2012); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997) and *DIAMOND* (Brandenburg, 2006); software used to prepare material for publication: *publCIF* (Westrip, 2010).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB6843).

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supplementary materials

Acta Cryst. (2012). E68, o2134 [doi:10.1107/S1600536812026189]

2-Methylsulfanyl-9H-1,3,4-thiadiazolo[2,3-b]quinazolin-9-one

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Comment

Quinazoline-4(3H)-one derivatives attract interest owing to their putative biological activity (El-Azab & ElTahir, 2012; El-Azab *et al.*, 2011). The title compound, 2-(methylthio)-5H-[1,3,4]thiadiazolo[2,3-b]quinazolin-5-one (I), has been synthesized previously and evaluated for its anti-microbial activity (El-Azab, 2007). Herein, we describe its crystal structure determination.

The 16 non-hydrogen atoms in (I), Fig. 1, are planar with the r.m.s. deviation being 0.037 Å. The maximum deviations from the least-squares plane are 0.068 (1) Å for the ketone-O1 atom and -0.065 (2) Å for the methyl-C10 atom. The *S*-bound methyl group is *syn* to the ketone-O1 atom.

In the crystal packing, centrosymmetrically related molecules are connected by C—H···O interactions between the ketone-O and methyl-H atoms, Table 1, *via* a 16-membered {···HCSCN₂CO₂}₂ synthon, Fig. 2. The dimeric aggregates are connected into a linear supramolecular chain along the *b* axis *via* π — π interactions occurring between the five-membered and benzene rings [inter-centroid distance = 3.6123 (9) Å, angle of inclination = 2.09 (7)° for symmetry operation: *x*, 1 + *y*, *z*]. The chains assemble into layers in the *bc* plane *via* S···S interactions involving the endocyclic-S1 atoms whereby each S1 atom forms two such interactions [S1···S1^{*i*} = 3.4607 (6) Å for symmetry operation *i*: 2 - *x*, 2 - *y*, 1 - *z*; and S1···S1^{*ii*} = 3.4792 (6) Å for *ii*: 2 - *x*, 1 - *y*, 1 - *z*]. Layers stack along the *a* axis without specific interactions between them, Fig. 3.

Experimental

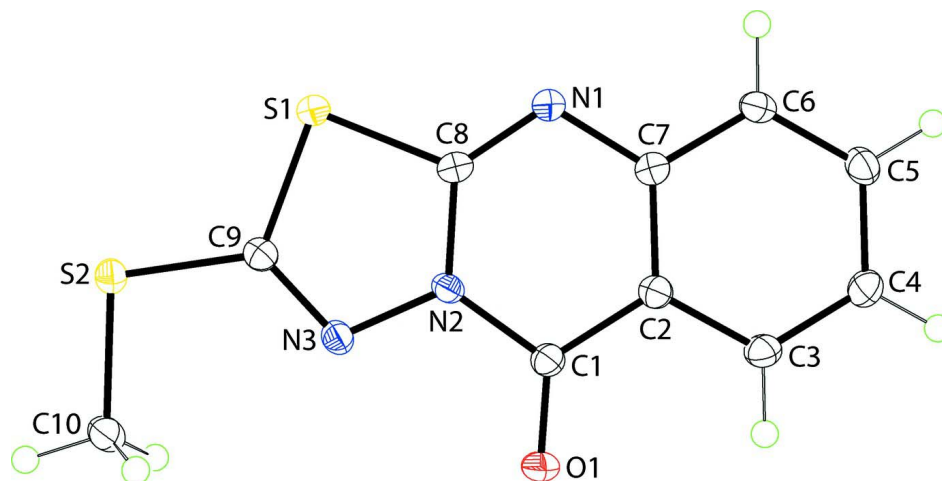
A mixture of 2-mercapto-5H-[1,3,4]thiadiazolo[2,3-b]quinazolin-5-one (470 mg, 2 mmol) and methyl iodide (2.1 mmol) in acetone (10 ml) containing anhydrous potassium carbonate (300 mg) was stirred at room temperature for 12 h. The reaction mixture was filtered, the solvent removed under reduced pressure and the solid obtained was dried and recrystallized from ethanol. Yield 88%. ¹H NMR (CDCl₃): δ 8.42 (d, 1H, *J* = 7.5 Hz), 7.79 (t, 1H, *J* = 7.0 Hz), 7.63 (d, 1H, *J* = 8.0 Hz), 7.49 (t, 1H, *J* = 7.0 Hz), 2.84 (s, 3H) p.p.m.. ¹³C NMR (CDCl₃): δ = 15.3, 118.9, 126.2, 127.6, 134.8, 147.2, 156.2, 157.1, 158.5 p.p.m..

Refinement

Carbon-bound H-atoms were placed in calculated positions [C—H = 0.95 to 0.98 Å, $U_{\text{iso}}(\text{H}) = 1.2\text{--}1.5U_{\text{eq}}(\text{C})$] and were included in the refinement in the riding model approximation.

Computing details

Data collection: *CrysAlis PRO* (Agilent, 2012); cell refinement: *CrysAlis PRO* (Agilent, 2012); data reduction: *CrysAlis PRO* (Agilent, 2012); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997) and *DIAMOND* (Brandenburg, 2006); software used to prepare material for publication: *publCIF* (Westrip, 2010).

**Figure 1**

The molecular structure of (I) showing displacement ellipsoids at the 50% probability level.

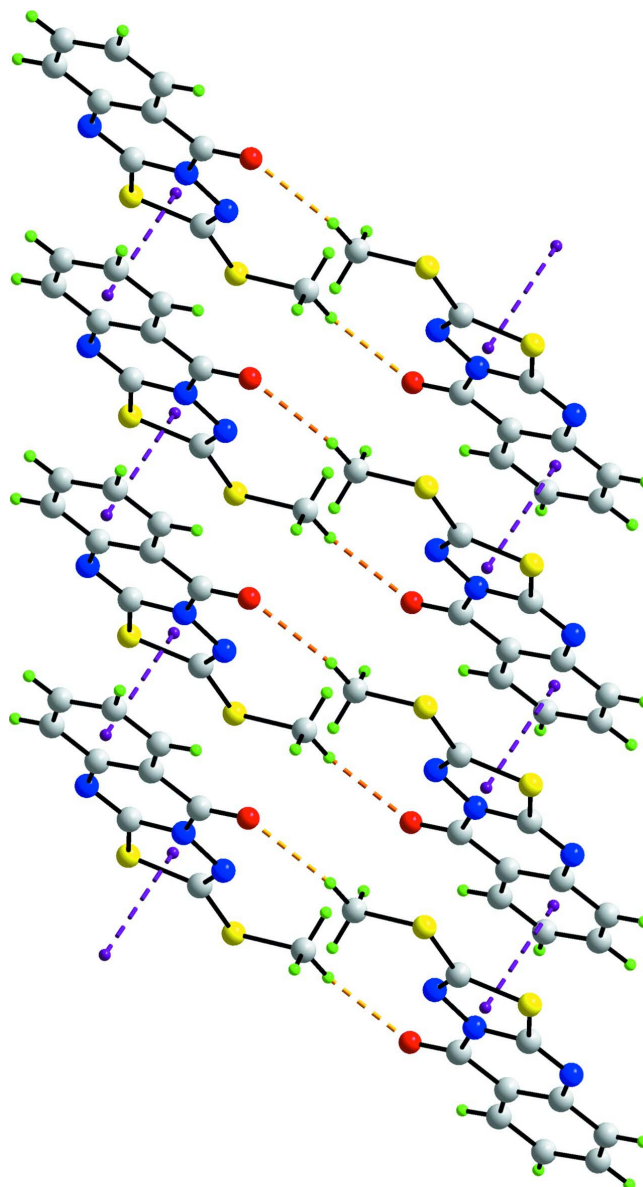
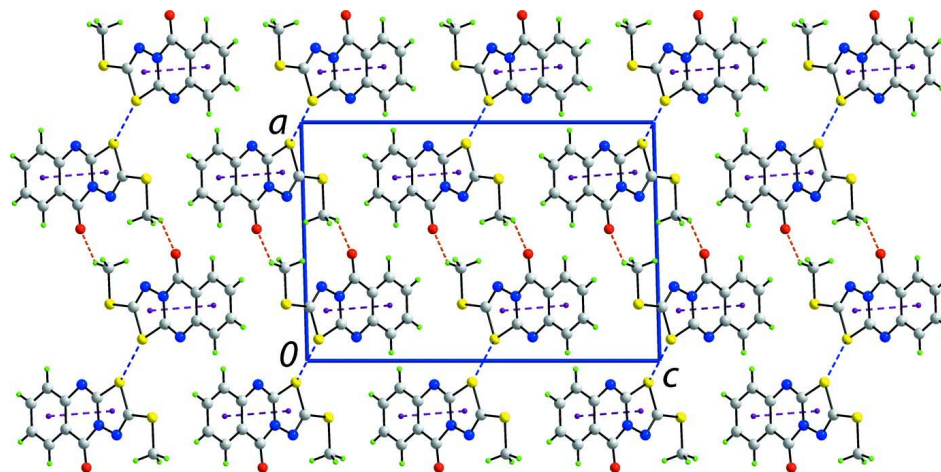


Figure 2

A view of the linear supramolecular chain along the *b* axis in (I). The C—H···O and π — π interactions are shown as orange and purple dashed lines respectively.


Figure 3

A view in projection down the b axis of the unit-cell contents for (I). The C—H \cdots O, π — π and S \cdots S interactions are shown as orange, purple and blue dashed lines respectively.

2-Methylsulfanyl-9H-1,3,4-thiadiazolo[2,3-*b*]quinazolin-9-one

Crystal data

$C_{10}H_7N_3OS_2$

$M_r = 249.31$

Monoclinic, $P2_1/c$

Hall symbol: $-P\ 2_1/c$

$a = 11.8193\ (4)\ \text{\AA}$

$b = 4.9841\ (2)\ \text{\AA}$

$c = 17.4985\ (6)\ \text{\AA}$

$\beta = 91.453\ (3)^\circ$

$V = 1030.48\ (6)\ \text{\AA}^3$

$Z = 4$

$F(000) = 512$

$D_x = 1.607\ \text{Mg m}^{-3}$

Cu $K\alpha$ radiation, $\lambda = 1.54184\ \text{\AA}$

Cell parameters from 2166 reflections

$\theta = 3.7\text{--}76.5^\circ$

$\mu = 4.53\ \text{mm}^{-1}$

$T = 100\ \text{K}$

Prism, colourless

$0.30 \times 0.10 \times 0.03\ \text{mm}$

Data collection

Agilent SuperNova Dual

diffractometer with an Atlas detector

Radiation source: SuperNova (Cu) X-ray

Source

Mirror monochromator

Detector resolution: $10.4041\ \text{pixels mm}^{-1}$

ω scan

Absorption correction: multi-scan

(*CrysAlis PRO*; Agilent, 2012)

$T_{\min} = 0.344$, $T_{\max} = 0.876$

3781 measured reflections

2110 independent reflections

1937 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.018$

$\theta_{\max} = 76.7^\circ$, $\theta_{\min} = 3.7^\circ$

$h = -12 \rightarrow 14$

$k = -5 \rightarrow 6$

$l = -15 \rightarrow 21$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.029$

$wR(F^2) = 0.082$

$S = 1.09$

2110 reflections

145 parameters

0 restraints

Primary atom site location: structure-invariant

direct methods

Secondary atom site location: difference Fourier

map

Hydrogen site location: inferred from

neighbouring sites

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0457P)^2 + 0.3686P]$$

where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$

$$\Delta\rho_{\max} = 0.36 \text{ e } \text{\AA}^{-3}$$

$$\Delta\rho_{\min} = -0.27 \text{ e } \text{\AA}^{-3}$$

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.90968 (3)	0.75129 (8)	0.46619 (2)	0.01702 (12)
S2	0.76245 (3)	1.12015 (8)	0.55613 (2)	0.01660 (12)
N1	0.90012 (11)	0.3645 (3)	0.35873 (8)	0.0160 (3)
N2	0.72981 (11)	0.5607 (3)	0.40394 (7)	0.0145 (3)
N3	0.68915 (11)	0.7516 (3)	0.45394 (7)	0.0160 (3)
O1	0.55197 (9)	0.4218 (3)	0.36566 (7)	0.0217 (3)
C1	0.65417 (13)	0.4007 (3)	0.36022 (9)	0.0159 (3)
C2	0.71312 (13)	0.2152 (3)	0.31034 (9)	0.0153 (3)
C3	0.64945 (13)	0.0467 (4)	0.26154 (9)	0.0183 (3)
H3	0.5691	0.0533	0.2616	0.022*
C4	0.70332 (14)	-0.1282 (4)	0.21351 (9)	0.0196 (3)
H4	0.6602	-0.2429	0.1806	0.024*
C5	0.82196 (14)	-0.1368 (3)	0.21325 (9)	0.0192 (3)
H5	0.8589	-0.2560	0.1796	0.023*
C6	0.88539 (13)	0.0261 (3)	0.26135 (9)	0.0179 (3)
H6	0.9657	0.0173	0.2609	0.021*
C7	0.83228 (13)	0.2050 (3)	0.31091 (9)	0.0149 (3)
C8	0.84595 (12)	0.5289 (3)	0.40102 (8)	0.0149 (3)
C9	0.77423 (13)	0.8644 (3)	0.48930 (9)	0.0152 (3)
C10	0.61018 (14)	1.1556 (4)	0.55743 (10)	0.0211 (3)
H10A	0.5903	1.2965	0.5938	0.032*
H10B	0.5762	0.9854	0.5730	0.032*
H10C	0.5815	1.2041	0.5062	0.032*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.01263 (19)	0.0180 (2)	0.0204 (2)	-0.00013 (13)	-0.00028 (14)	-0.00389 (14)
S2	0.0170 (2)	0.0176 (2)	0.0151 (2)	0.00105 (14)	-0.00022 (14)	-0.00248 (14)
N1	0.0136 (6)	0.0176 (7)	0.0167 (6)	0.0001 (5)	0.0003 (5)	-0.0019 (5)
N2	0.0124 (6)	0.0179 (7)	0.0133 (6)	0.0031 (5)	0.0007 (5)	-0.0011 (5)
N3	0.0161 (6)	0.0180 (7)	0.0139 (6)	0.0031 (5)	0.0001 (5)	-0.0018 (5)
O1	0.0118 (5)	0.0294 (7)	0.0239 (6)	0.0024 (5)	-0.0007 (4)	-0.0067 (5)
C1	0.0148 (7)	0.0191 (7)	0.0138 (7)	0.0014 (6)	-0.0011 (6)	-0.0006 (6)
C2	0.0151 (7)	0.0176 (7)	0.0131 (7)	0.0011 (6)	-0.0001 (5)	0.0006 (6)
C3	0.0146 (7)	0.0227 (8)	0.0176 (7)	0.0013 (7)	-0.0018 (6)	-0.0009 (6)
C4	0.0190 (8)	0.0219 (8)	0.0177 (8)	0.0007 (6)	-0.0041 (6)	-0.0031 (6)
C5	0.0204 (8)	0.0207 (8)	0.0166 (7)	0.0037 (6)	0.0012 (6)	-0.0032 (6)
C6	0.0135 (7)	0.0206 (8)	0.0195 (7)	0.0022 (6)	0.0013 (6)	-0.0013 (6)
C7	0.0149 (7)	0.0160 (7)	0.0139 (7)	-0.0004 (6)	0.0001 (5)	0.0017 (6)
C8	0.0127 (7)	0.0159 (7)	0.0160 (7)	-0.0008 (6)	-0.0005 (5)	0.0023 (6)
C9	0.0155 (7)	0.0160 (7)	0.0141 (7)	0.0019 (6)	0.0005 (5)	0.0014 (6)
C10	0.0178 (7)	0.0261 (9)	0.0194 (8)	0.0048 (7)	0.0015 (6)	-0.0032 (7)

Geometric parameters (Å, °)

S1—C8	1.7473 (16)	C2—C7	1.409 (2)
S1—C9	1.7542 (16)	C3—C4	1.378 (2)
S2—C9	1.7378 (16)	C3—H3	0.9500
S2—C10	1.8091 (17)	C4—C5	1.403 (2)
N1—C8	1.286 (2)	C4—H4	0.9500
N1—C7	1.393 (2)	C5—C6	1.378 (2)
N2—C8	1.3840 (18)	C5—H5	0.9500
N2—N3	1.3866 (18)	C6—C7	1.403 (2)
N2—C1	1.409 (2)	C6—H6	0.9500
N3—C9	1.296 (2)	C10—H10A	0.9800
O1—C1	1.2186 (19)	C10—H10B	0.9800
C1—C2	1.461 (2)	C10—H10C	0.9800
C2—C3	1.403 (2)		
C8—S1—C9	88.48 (7)	C6—C5—H5	119.7
C9—S2—C10	100.19 (8)	C4—C5—H5	119.7
C8—N1—C7	114.96 (13)	C5—C6—C7	120.47 (14)
C8—N2—N3	117.51 (13)	C5—C6—H6	119.8
C8—N2—C1	122.09 (13)	C7—C6—H6	119.8
N3—N2—C1	120.37 (12)	N1—C7—C6	118.30 (14)
C9—N3—N2	108.78 (13)	N1—C7—C2	122.93 (14)
O1—C1—N2	121.69 (14)	C6—C7—C2	118.77 (14)
O1—C1—C2	126.16 (15)	N1—C8—N2	127.10 (14)
N2—C1—C2	112.15 (13)	N1—C8—S1	124.55 (12)
C3—C2—C7	120.23 (14)	N2—C8—S1	108.35 (11)
C3—C2—C1	119.08 (14)	N3—C9—S2	124.42 (12)
C7—C2—C1	120.69 (14)	N3—C9—S1	116.88 (12)
C4—C3—C2	120.06 (15)	S2—C9—S1	118.69 (9)
C4—C3—H3	120.0	S2—C10—H10A	109.5
C2—C3—H3	120.0	S2—C10—H10B	109.5
C3—C4—C5	119.90 (15)	H10A—C10—H10B	109.5
C3—C4—H4	120.0	S2—C10—H10C	109.5
C5—C4—H4	120.0	H10A—C10—H10C	109.5
C6—C5—C4	120.56 (15)	H10B—C10—H10C	109.5
C8—N2—N3—C9	-0.24 (19)	C3—C2—C7—N1	-179.16 (15)
C1—N2—N3—C9	177.60 (13)	C1—C2—C7—N1	0.9 (2)
C8—N2—C1—O1	176.66 (15)	C3—C2—C7—C6	0.7 (2)
N3—N2—C1—O1	-1.1 (2)	C1—C2—C7—C6	-179.20 (15)
C8—N2—C1—C2	-3.1 (2)	C7—N1—C8—N2	0.4 (2)
N3—N2—C1—C2	179.13 (13)	C7—N1—C8—S1	179.87 (11)
O1—C1—C2—C3	1.9 (3)	N3—N2—C8—N1	-179.87 (15)
N2—C1—C2—C3	-178.30 (14)	C1—N2—C8—N1	2.3 (2)
O1—C1—C2—C7	-178.17 (16)	N3—N2—C8—S1	0.61 (17)
N2—C1—C2—C7	1.6 (2)	C1—N2—C8—S1	-177.18 (12)
C7—C2—C3—C4	-0.5 (2)	C9—S1—C8—N1	179.88 (15)
C1—C2—C3—C4	179.44 (15)	C9—S1—C8—N2	-0.59 (11)
C2—C3—C4—C5	-0.3 (3)	N2—N3—C9—S2	178.77 (11)

C3—C4—C5—C6	0.8 (3)	N2—N3—C9—S1	-0.27 (17)
C4—C5—C6—C7	-0.5 (3)	C10—S2—C9—N3	0.76 (16)
C8—N1—C7—C6	178.13 (15)	C10—S2—C9—S1	179.79 (10)
C8—N1—C7—C2	-2.0 (2)	C8—S1—C9—N3	0.53 (13)
C5—C6—C7—N1	179.67 (15)	C8—S1—C9—S2	-178.57 (10)
C5—C6—C7—C2	-0.2 (2)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C10—H10 <i>A</i> \cdots O1 ⁱ	0.98	2.32	3.170 (2)	145

Symmetry code: (i) $-x+1, -y+2, -z+1$.